Nitrogen isotopes in ophiolitic metagabbros: A re-evaluation of modern nitrogen fluxes in subduction zones and implication for the early Earth atmosphere

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Received 17 May 2011; accepted in revised form 29 September 2011; available online 6 October 2011

Abstract

Nitrogen contents and isotope compositions together with major and trace element concentrations were determined in a sequence of metagabbros from the western Alps (Europe) in order to constrain the evolution and behavior of N during hydrothermal alteration on the seafloor and progressive dehydration during subduction in a cold slab environment (8 °C/km). The rocks investigated include: (i) low-strain metagabbros that equilibrated under greenschist to amphibolite facies (Chenaillet Massif), blueschist facies (Queyras region) and eclogite facies (Monviso massif) conditions and (ii) highly-strained mylonites and associated eclogitic veins from the Monviso Massif. In all samples, nitrogen (2.6–55 ppm) occurs as bound ammonium (NH₄⁺) substituting for K or Na–Ca in minerals. Cu concentrations show a large variation, from 73.2 to 6.4 ppm, and are used as an index of hydrothermal alteration on the seafloor because of Cu fluid-mobility at relatively high temperature (>300 °C). In low-strain metagabbros, δ¹⁵N values of +0.8‰ to +8.1‰ are negatively correlated with Cu concentrations. Eclogitic mylonites and veins display Cu concentrations lower than 11 ppm and show a δ¹⁵N–Cu relationship that does not match the δ¹⁵N–Cu correlation found in low-strain rocks. This δ¹⁵N–Cu correlation preserved in low-strain rocks is best interpreted by leaching of Cu–N compounds, possibly of the form Cu(NH₃)₂²⁺, during hydrothermal alteration. Recognition that the different types of low-strain metagabbros show the same δ¹⁵N–Cu correlation indicates that fluid release during subduction zone metamorphism did not modify the original N and Cu contents of the parent hydrothermally-altered metagabbros. In contrast, the low Cu content present in eclogitic veins and mylonites implies that ductile deformation and veining were accomplished either by a loss of copper or that externally-derived nitrogen was added to the system.

We estimate the global annual flux of N subducted by metagabbros as 4.2 (±2.0) × 10¹¹ g/yr. This value is about half that of sedimentary rocks and that gabbros carry a significant portion of the subducted nitrogen. The net budget between subducted N and that outgassed at volcanic arcs indicates that ~80% of the subducted N is not recycled to the surface. On a global scale, the total amount of N buried to the mantle via subduction zones is estimated to be three times higher than that released from the mantle via mid-ocean ridges, and intraplate volcanoes and back-arc basins. This implies that N contained in Earth surface reservoirs, mainly in the atmosphere, is progressively transferred and sequestered into the mantle, with a net flux of ~9.6 × 10¹¹ g/yr. Assuming a constant flux of subducted N over the Earth’s history indicates that an amount equivalent to the present atmospheric N may have been sequestered into the silicate Earth over a period of 4 billion years. © 2011 Elsevier Ltd. All rights reserved.

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1. INTRODUCTION: A BRIEF REVIEW OF NITROGEN CHEMICAL GEO DYNAMICS

Nitrogen is a volatile element that can help better addressing the origin and fate of the Earth’s atmosphere. This is because, unlike carbon or water, the nitrogen geodynamic cycle during mantle-surface exchanges is far from steady state. Nitrogen unsteady-state (also referred to as nitrogen disequilibrium; e.g. Boyd and Pillinger, 1994) is best illustrated by the contrasting isotope compositions of present-day degassed and recycled nitrogen, having negative and positive $\delta^{15}\text{N}$-values of $\sim-5_{\text{ppm}}$ (where $\delta^{15}\text{N} = [^{15}\text{N}/^{14}\text{N}_{\text{sample}}]/[^{15}\text{N}/^{14}\text{N}_{\text{Air}}] - 1 \times 1000$; Javoy et al., 1984; Marty and Humbert, 1997) and $>+2_{\text{ppm}}$ (for a review of N-isotope data in sediments, see Thomazo et al., 2009; for a discussion on the fate of nitrogen during subduction, see Bebout and Fogel, 1992; Busigny et al., 2003a,b). This predicts secular evolutions of the mantle (increasing $\delta^{15}\text{N}$), crust and atmosphere reservoirs (decreasing $\delta^{15}\text{N}$) through time (Javoy, 1998; Tolstikhin and Marty, 1998).

Nitrogen isotope composition of the early Earth mantle would be recorded in a diamond with the lowest $\delta^{15}\text{N}$ value at $-25_{\text{ppm}}$ (Cartigny et al., 1997). Very positive $\delta^{15}\text{N}$-values (up to $24_{\text{ppm}}$) in both Archean sediments and metamorphic rocks have been interpreted to reflect the secular evolution of the atmosphere (Jia and Kerrich, 2004a) but these values contrast with other data sets illustrating very little nitrogen isotope variations of the atmosphere through time (Sano and Pillinger, 1990; Beaumont and Robert, 1999). A potential explanation might be found in the recent study of $\sim2.7$ Gy sediments illustrating high $\delta^{15}\text{N}$-values, up to $+50_{\text{ppm}}$, suggested to result from ammonia oxidation (Thomazo et al., 2010) rather than secular evolution of N-isotope composition in the atmosphere.

Accurately modeling the origin of the atmosphere, its chemical evolution and fate heavily relies on our knowledge of (i) N concentrations and isotope compositions in major Earth reservoirs together with (ii) fluxes between these reservoirs (e.g. Javoy, 1998; Tolstikhin and Marty, 1998; Marty and Dauphas, 2003; Jia and Kerrich, 2004a; Goldblatt et al., 2009). There is still some disagreement about present-day mantle N-contents (see Marty, 1995; Cartigny et al., 2001a, for discussion) and very few data are presently available for crustal nitrogen (Wedepohl, 1995; Boyd and Philippot, 1998; Palya et al., 2011).

Degassing and recycling fluxes are comparatively better established. Unlike noble gases which are accumulated within the Earth’s atmosphere and not subducted into the mantle, nitrogen is massively recycled with a subducted/degassed ratio of $\sim10$ (see below). This is inferred from a present-day nitrogen degassing flux at mid-ocean ridges of $\sim0.7 \times 10^{11}$ g/yr, which is consistent with both N/He/Ar, C/N (Marty, 1995; Javoy and Pineau, 1991) and re-evaluated mantle C-flux and C/Nb-systematics (Saal et al., 2002; Cartigny et al., 2008). Secondary output fluxes are brought by back-arc basins and intraplate volcanism, which contribute, respectively, $7.8 \times 10^7$ and $5.7 \times 10^7$ g/yr (Sano et al., 2001). The flux released by arc volcanism is difficult to establish owing to the high variability of fluxes in time and space for fumaroles and volcanic gases. The value adopted here, $2.8 \times 10^{11}$ g/yr, is taken from Hilton et al. (2002) as it consists of the largest database (>700 samples), contrasting with the 11 samples used for another lower estimate ($8.9 \times 10^9$ g/yr; Sano et al., 2001).

The flux of N input in subduction zones can be determined from analyses of the various components of the subducting oceanic lithosphere, including sediments, basalts, gabbros and serpentinized peridotites. Analyses of metasediments in paleo-subduction zones showed that N behavior in subducting sediments is mainly controlled by the geothermal gradient (Bebout et al., 1999; Busigny et al., 2003a). Subduction along “warm” geothermal gradient ($\sim15 \degree C/km$) induces significant release of N from rocks to fluids (Haenel et al., 1986; Bebout and Fogel, 1992; Mingram and Bräuer, 2001), associated with N isotope fractionation (i.e. loss of $^{15}\text{N}$-depleted components). In contrast, metasediments subducted along “cold” geothermal gradient ($\sim8 \degree C/km$) do not show any significant loss of N into fluids (Busigny et al., 2003a; Pitcairn et al., 2005). This observation reflects the high stability of N as ammonium ($\text{NH}_4^+$) in high-pressure white micas. The global flux of N carried by subducting metasediments was estimated to $7.6 \times 10^{10}$ g/yr based on the relations between N and K abundances (Busigny et al., 2003a). This flux is three times higher than the mean flux output from arc volcanism, implying that N may be massively recycled to the mantle through subduction zones. Other fluxes, transferred by magmatic and ultramafic components of the subducting oceanic lithosphere, are still missing but may contribute to surprisingly large amount of N for efficient N burial to the deep mantle (Li et al., 2007; Halama et al., 2010). Two recent studies reported N content and isotope composition of serpentinized metaperidotites from paleo-subduction zones (Philippot et al., 2007; Halama et al., 2010) and found N content between 1.4 and 15 ppm N, with very positive $\delta^{15}\text{N}$ values (+4$_{\text{ppm}}$ to +15$_{\text{ppm}}$). While this may represent a significant portion of the total N input in subduction zones, the flux of N carried by serpentinites was not assessed because the mass flux of subducting serpentinites is largely unknown. Studies of basalts drilled on the ocean seafloor have shown that alteration by seawater produces an enrichment of N in the rock (up to 18 ppm N in bulk rock) due to fluid–rock interaction (Busigny et al., 2005a; Li et al., 2007). Similar N concentrations were found in high- and ultrahigh-pressure mafic eclogites, suggesting that N stored in altered oceanic crust can be deeply and efficiently subducted (Halama et al., 2010). Again, too few data are yet available to accurately establish the N-flux related to the upper part of the oceanic crust, but increasing amount of data illustrates that these appear non-negligible compared with sedimentary flux.

In this respect, large uncertainties remain for the gabbroic part as it is the only lithological component that has not been explored so far for N concentration and isotope composition. This is a significant gap in our knowledge of the global N budget since gabbros represent 2/3 of the total mass of subducting oceanic crust, the other 1/3 being basalts (e.g. Peacock, 1990). In the present work, we report N content and isotope composition of ophiolitic gabbros from Western Alps (Europe). These rocks were em-
placed in the oceanic crust and altered by seawater-derived fluids before being subducted to various P–T conditions (blueschist and eclogite facies up to 2 GPa and 500 °C). Non-subducted equivalents were also analyzed to differentiate chemical changes related to subduction metamorphism from those related to hydrothermal circulation. Major and trace elements data were analyzed in order to determine potential co-variations with N, and constrain N geochemical behavior. In particular, the degree of Cu depletion is applied as a tracer of alteration by high-temperature hydrothermal fluids (e.g. Seewald and Seyfried, 1990; Heft et al., 2008). This unique data set is used to understand (1) the parameters controlling N concentration and isotope composition in ophiolitic metagabbros, (2) the fate of N carried by gabbros in subduction zones, (3) the flux of N deeply subducted and recycled to the mantle. Finally, the flux of N carried by subducting metagabbros is compared to other input fluxes by metasediments and metabasalts, and implications for global N geodynamic cycle and for the atmosphere evolution are discussed.

### 2. GEOLOGICAL SETTING AND SAMPLE DESCRIPTION

The metagabbros analyzed in this study occur together with metabasalts and metaperidotites as dismembered meta-ophiolitic bodies embedded in pelagic metasediments that form the so-called “nappe of Schistes Lustres” of the Piemonte-Ligurian domain, western Alps (Europe; Fig. 1). These rocks experienced hydrothermal alteration on the seafloor and were in part, subsequently subducted under the Adria margin during the Late Cretaceous-Eocene in a relatively cold slab environment (~8 °C/km; Le Pichon et al., 1988; Monié and Philippot, 1989). Samples analyzed in this study include hydrothermally-altered metagabbros from the Chenailliet Massif that escaped subduction zone metamorphism (Mével et al., 1978; Manatschal et al., 2011), and blueschist- and eclogite-facies metagabbros from the Queyras region and Monviso Massif, respectively. Analyzed samples include low-strain rocks from Chenailliet, Queyras and Monviso (Fig. 2a and b) together with eclogite facies mylonites and veins from Monviso (Fig. 2c and d).

#### 2.1. Hydrothermally-altered Chenailliet metagabbros

Chenailliet metagabbros were used as reference material predating subduction zone metamorphism. Two samples previously investigated by Nadeau et al. (1993) for their oxygen and carbon isotope compositions were analyzed here (CH-129 and CH-80-02). These samples are undeformed metagabbros mainly composed of large crystals (mm to cm scale) of magmatic pyroxene (augite), plagioclase, apatite and ilmenite. These rocks have experienced oceanic hydrothermal alteration under greenschist- to amphibolite-facies conditions. Primary pyroxene has been partially replaced by actinolite, chlorite and green hornblende while some plagioclase has been partly pseudomorphed into epidote and calcite.

#### 2.2. Queyras metagabbros (blueschist facies)

Four low-strain metagabbros from the Queyras region were analyzed (09-02WR, G6004A, G6004B and G6004C). The samples are mainly composed of large crystals of glauconephane (>50 vol%) in association with lawsonite and plagioclase and minor amounts of epidote, sphenecomphane (<5 vol%). Few primary magmatic pyroxenes are preserved in all samples. They are occasionally associated with rare hornblendes inherited from the hydrothermal alteration stage predating blueschist-facies overprint (samples G6004B and G6004C). Minor occurrences of secondary actinolite and chlorite pseudomorphing blueschist-facies minerals indicate that late stage retrogression under greenschist-facies conditions was limited.

#### 2.3. Monviso metagabbros (eclogite facies)

The samples analyzed from the Monviso Massif have been subjected to extensive petrological, structural and geochemical investigations (Philippot, 1987, 1993; Philippot and Kienast, 1989; Philippot and Selverstone, 1991; Philippot and van Roermund, 1992; Nadeau et al., 1993). They belong to the Lago Superiore unit, which consists of a sequence of cumulates ranging from undifferentiated Mg–Cr metagabbros at the bottom, to more differentiated Fe–Ti metagabbros at the top (see Philippot and Kienast, 1989). In the sequence, the deformation is very heterogeneous. Strongly deformed domains with mylonitic to ultra-mylonitic textures surround low-strain domains, which preserved pre-subduction magmatic textures (Philippot and van Roermund, 1992; Nadeau et al., 1993). Highly strained domains are commonly cut across by omphacite- and garnet-bearing veins that formed during eclogite-facies metamorphism. P–T conditions estimates yield a peak pressure of about 2.5 GPa at about 550 ± 50 °C (Messiga et al., 1999). Seven well-preserved eclogitic metagabbros were selected for N isotope study. These are: two low-strain rocks (VS-1 and Vi-52), three mylonites (Vi–385M, Vi–262M and Vi–389M), and four eclogitic veins (VS–14V, Vi–79V, Vi–387V and Vi–385V). All samples contain the same omphacite (>50 vol%), garnet, rutile and apatite mineral assemblage. In addition, VS–14V contains minor talc; Vi–385V and Vi–389M, which represent a vein (V) and its adjacent mylonitic host (M), contain minor phengite (Philippot and Kienast, 1989). The low-strain Mg–Cr metagabbro (Vi-52) contains Mg–chloritoid, omphacite, spinel (chromite), talc, garnet and rutile. δ18O values of omphacite between +3.0‰ to +5.3‰ (average ~+3.5‰, Nadeau et al., 1993) were attributed to be inherited from hydrothermal alteration at high temperature without subsequent re-equilibration during subduction-zone metamorphism under blueschist- and eclogite-facies conditions (Philippot et al., 1998).

### 3. ANALYTICAL TECHNIQUES

The procedure used for the determination of nitrogen content and isotope composition in rock samples is an improved method from Kendall and Grim (1990). It was
Nitrogen isotopes in ophiolitic metagabbros

1 - Chenaillat Massif
   (amphibolite facies; hydrothermal)
2 - Queyras Valley
   (blueschist facies)
3 - Monviso Massif
   (eclogite facies)

Fig. 1. Schematic map of the main tectonic units of the western Alps (modified after Agard et al., 2001). Internal crystalline massifs are from south to north: Dora Maira (DM), Grand Paradiso (GP), Monte Rosa (MR). Rectangles represent sampling zones from 1 to 3, corresponding to increasing metamorphic conditions.

thoroughly described in previous contributions (Busigny et al., 2005b; Ader et al., 2006) and is only summarized herein. Nitrogen was extracted from bulk-rock powders by combustion in quartz tubes sealed under vacuum with purified reactants (Cu, CuO and CaO). It was then quantified as N$_2$ by capacitance manometry in ultra-high vacuum line and its isotopic analysis was performed on a triple-collector static vacuum mass spectrometer. The N blank of the entire procedure is low (0.65 ± 0.30 nmol N), with mean $\delta^{15}$N value of $-3.7 \pm 2.7 \%$ (2σ n = 40; Busigny et al., 2005b). The precision for nitrogen contents and $\delta^{15}$N values measured in our samples is better than ±8% and ±0.5% respectively.

Whole-rock major and trace elements concentrations were measured using ICP-AES and ICP-MS at the Service d’Analyses des Roches et des Minéraux (SARM) of the Centre de Recherches Pétrographiques et Géochimiques (CRPG) of Vandoeuvre, France. Analytical precisions and detection limits of major and trace element concentrations are available at http://helium.crpg.cnrs-nancy.fr/SARM/pages/roches.html and are typically ±10%.

For two samples (Vi–385V and Vi–79V), some phengites were identified and separated from their host-rock. These single phengite grains were analyzed by infrared spectroscopy using the method described in Busigny et al. (2003b,
in order to detect and, if present, quantify ammonium, \(\text{NH}_4^+\) (substituting for \(\text{K}^+\)). Infrared spectra all show large absorption bands typical of \(\text{NH}_4\)-bending at \(\sim 1430 \text{ cm}^{-1}\) and \(\text{NH}_4\)-stretching between 2750 and 3400 \(\text{cm}^{-1}\).

4. RESULTS

4.1. Major and trace element data

Major and trace element concentrations (Tables 1 and 2) show that the metagabbros studied represent variable degrees of magmatic differentiation. \(\text{Mg}^\#\) (i.e. \(100 \times \text{MgO}/[\text{MgO} + \text{FeO} + \text{Fe}_2\text{O}_3]\), where \(\text{MgO}\), \(\text{FeO}\) and \(\text{Fe}_2\text{O}_3\) are molar abundances) is usually applied to trace magmatic differentiation because \(\text{Mg}/\text{Fe}\) ratio progressively decreases with increasing differentiation. However, our samples experienced fluid–rock interactions during hydrothermal alteration and subduction processes. Since \(\text{Mg}\) and \(\text{Fe}\) can be affected by fluid–rock exchanges (Alt, 1995; Alt and Teagle, 2003; Staudigel, 2003), \(\text{Fe}/\text{Mg}\) may have been modified and \(\text{Mg}^\#\) may not simply depend upon magmatic differentiation. In order to test this hypothesis, we have reported \(\text{Mg}^\#\) as a function of concentrations of elements, which are immobile during fluid–rock exchange (e.g. HFSE, REE; Staudigel, 2003; Spandler et al., 2004). Incompatible and fluid-immobile elements like \(\text{Zr}, \text{Hf}, \text{Nb}\) (HFSE) and \(\text{Lu}\) (REE) show an increasing concentration with decreasing \(\text{Mg}^\#\) (Fig. 3). On the contrary, \(\text{Cr}\), which is a compatible element, shows a progressive decrease over several orders of magnitude (from 7.3 to 17,900 ppm) with decreasing \(\text{Mg}^\#\) (Fig. 3). These correlations indicate that \(\text{Mg}^\#\) was only moderately affected by hydrothermalism and subduction (e.g. Staudigel, 2003). Thus major elements primarily reflect magmatic processes at mid-ocean ridges. Other major elements such as \(\text{Fe}, \text{Mn}\), and \(\text{Ti}\) show increasing concentrations with decreasing \(\text{Mg}^\#\) (Fig. 3), reflecting the well-known incompatible behavior of \(\text{Fe}, \text{Mn}\) and \(\text{Ti}\) relative to \(\text{Mg}\). \(\text{Cu}\) content of low-strain metagabbros show a wide range of variation (from 6.4 to 73.2 ppm) and tends to correlate with \(\text{Mg}^\#\) (except for the \(\text{Mg}\)-rich metagabbrs). Veins and mylonites have low \(\text{Cu}\) concentrations (less than 11 ppm for all samples but one vein with 27 ppm). Aluminum concentration roughly decreases, reflecting its compatible behavior during plagioclase crystallization. These chemical features can be used to distinguish two groups of gabbros, being either \(\text{Mg–Cr}\) rich or \(\text{Fe–Ti}\) rich rocks (Lombardo et al., 1978). Since the gabbros are cumulates, they could not necessarily fall on differentiation trends. For example, some rocks could contain intercumulus melt and be enriched in incompatible elements, or others could be depleted in incompatible elements (e.g. Alt and Anderson, 1991). However, the present data generally show reasonable igneous trends while some variability may be explained by the fact that the rocks are cumulates.

4.2. Nitrogen concentration and isotope composition

Tables 3 and 4 report \(\text{N}\) concentrations and isotope compositions of whole-rock samples for low-strain metagabbros, mylonites and veins, respectively. All but one sample were analyzed at least twice and show a very good reproducibility (within ±0.5\%). Among the four
Nitrogen isopes in ophiolitic metagabbros

Table 1
Major and trace element concentrations in Alpine undeformed and low-strain metagabbros.

<table>
<thead>
<tr>
<th>Samples</th>
<th>CH-129</th>
<th>CH-80-02</th>
<th>G6004A</th>
<th>G6004B</th>
<th>G6004C</th>
<th>09-02WR</th>
<th>VS-1</th>
<th>Vi-52</th>
</tr>
</thead>
<tbody>
<tr>
<td>SiO₂</td>
<td>41.76</td>
<td>34.69</td>
<td>48.27</td>
<td>47.05</td>
<td>50.07</td>
<td>53.43</td>
<td>45.44</td>
<td>43.60</td>
</tr>
<tr>
<td>Al₂O₃</td>
<td>11.90</td>
<td>8.69</td>
<td>19.86</td>
<td>11.37</td>
<td>8.48</td>
<td>13.16</td>
<td>12.72</td>
<td>20.02</td>
</tr>
<tr>
<td>Fe₂O₃</td>
<td>16.90</td>
<td>21.61</td>
<td>7.31</td>
<td>16.73</td>
<td>16.11</td>
<td>9.06</td>
<td>14.93</td>
<td>4.61</td>
</tr>
<tr>
<td>MnO</td>
<td>0.29</td>
<td>0.37</td>
<td>0.13</td>
<td>0.27</td>
<td>0.19</td>
<td>0.14</td>
<td>0.21</td>
<td>0.05</td>
</tr>
<tr>
<td>MgO</td>
<td>4.73</td>
<td>5.20</td>
<td>3.31</td>
<td>5.91</td>
<td>5.06</td>
<td>9.48</td>
<td>6.34</td>
<td>13.44</td>
</tr>
<tr>
<td>CaO</td>
<td>11.69</td>
<td>14.66</td>
<td>9.57</td>
<td>7.08</td>
<td>5.64</td>
<td>4.93</td>
<td>10.66</td>
<td>9.91</td>
</tr>
<tr>
<td>Na₂O</td>
<td>3.66</td>
<td>2.78</td>
<td>4.87</td>
<td>4.78</td>
<td>5.75</td>
<td>5.66</td>
<td>3.91</td>
<td>2.40</td>
</tr>
<tr>
<td>K₂O</td>
<td>0.05</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>0.02</td>
<td></td>
</tr>
<tr>
<td>TiO₂</td>
<td>2.96</td>
<td>6.00</td>
<td>0.70</td>
<td>4.25</td>
<td>6.28</td>
<td>0.50</td>
<td>6.40</td>
<td>0.17</td>
</tr>
<tr>
<td>P₂O₅</td>
<td>4.10</td>
<td>5.49</td>
<td>0.12</td>
<td>0.64</td>
<td>0.04</td>
<td>0.09</td>
<td>0.05</td>
<td></td>
</tr>
<tr>
<td>LOI</td>
<td>1.49</td>
<td>0.54</td>
<td>4.39</td>
<td>2.32</td>
<td>0.90</td>
<td>3.23</td>
<td>0.83</td>
<td>3.59</td>
</tr>
<tr>
<td>Total</td>
<td>99.48</td>
<td>100.08</td>
<td>98.53</td>
<td>100.40</td>
<td>98.48</td>
<td>99.63</td>
<td>99.87</td>
<td>97.84</td>
</tr>
</tbody>
</table>

Cesium: 82.6 82.8 3.4 17.1 1.8 1.8 1.2
Carbon: 28.9 48.9 20.4 41.4 47.4 32.4 22.1 48.8
Nitrogen: 12.6 7.3 30.3 10.6 9.8 129 113 17,900
Sodium: – – 0.3 – – – – –
Calcium: 49.3 73.2 33.6 45.4 41.3 6.4 22.5 60.3
Magnesium: 28.3 29.4 2.3 10.8 2.5 1.8 3.7 –
Strontium: 13.5 13.8 1.4 5.9 1.7 1.1 2.3 –
Barium: 7.5 7.5 0.9 2.4 0.7 0.6 0.3 –
Radium: 17.3 14.4 20.7 20.1 12.1 12.3 – 13.0
Strontium: 33.6 35.2 2.0 10.0 1.7 1.5 1.2 –
Germanium: 1.6 1.8 1.8 1.5 2.1 1.0 – 1.0
Hafnium: 1.2 1.4 0.9 2.8 1.4 0.6 2.1 0.2
Holmium: 5.5 5.6 0.5 2.2 0.5 0.4 1.0 –
Indium: 0.1 0.1 0.1 0.2 0.1 0.1 0.1 –
Lanthane: 24.9 23.8 1.9 4.9 0.6 0.5 0.3 –
Lutetium: 1.3 1.3 0.2 0.8 0.3 0.2 0.4 –
Mendelevium: 0.9 0.8 0.6 0.9 0.4 – 0.6 –
Neodymium: 2.4 2.6 0.3 2.6 2.3 0.1 2.5 0.1
Nioptride: 91.4 92.9 4.1 21.0 2.5 2.3 1.4 –
Palladium: 27.5 40.8 99.4 13.5 75.8 72.2 46.3 477
Platinum: 4.2 – 8.2 6.1 6.6 – – –
Praseodymium: 16.1 15.9 0.8 3.5 0.4 0.4 0.2 –
Prometheum: – – 4.2 0.8 – 0.6 2.9 –
Smierium: 27.9 29.0 1.4 7.6 1.2 1.0 0.7 –
Samarium: 3.9 1.4 0.6 1.1 2.2 0.4 – 0.5
Selenium: 175 131 123 87.2 34.4 101 36.8 14.3
Thulium: 0.2 0.3 – 0.2 0.2 – – –
Thulium: 5.0 5.1 0.2 1.7 0.4 0.3 0.4 –
Thulium: 0.1 0.1 0.1 0.1 0.0 – – –
Thulium: 1.7 1.7 0.2 0.8 0.3 0.2 0.4 –
Uranium: 0.1 0.1 0.1 – 0.1 – – –
Vanadium: 69.5 381 256 447 692 166 699 102
Wolfram: 2.8 0.9 0.6 1.0 0.5 0.3 1.6 –
Yttrium: 170 173 13.5 59.4 15.4 10.4 24.2 0.2
Yttrium: 9.7 9.5 1.4 5.4 2.1 1.0 2.2 0.0
Zinc: 118 139 60.5 131 114 56.1 86.5 90.5
Zirconium: 47.3 57.6 32.7 106 55.1 17.3 80.8 6.8

Major and trace element concentrations are, respectively, in wt% and ppm.

high-pressure veins studied, two contain minor amounts of phengite (Vi–385V and Vi–79V). Estimates of the phengite NH₄⁺ concentrations by infrared spectroscopy yielded 584 ± 37 ppm (2σ, n = 5) and 1046 ± 70 ppm (2σ, n = 3) for the two samples, respectively.

Nitrogen amounts for mass spectrometer measurements range from 7 to 124 nmol N, which are more than 10 times higher than the average blank. With the exception of one eclogitic vein (VS–14V) that contains 55 ppm nitrogen (Table 4), whole-rock N contents range between 2.6 and 28 ppm. The N content of low-strain metagabbros shows a rough positive correlation with the sum of CaO + Na₂O concentrations (Fig. 4), which is used as a proxy of the amount of Ca–Na-bearing crystals in the samples (feldspars or augite in gabbros; sericite, zeolite, hornblende in hydrothermally-altered metagabbros; clinopyroxene in eclogitic metagabbros). δ¹⁵N values vary from +0.8‰ to +8.1‰, a range strikingly similar to that of HP/UHP eclogites.
Nitrogen concentration tends to decrease with increasing $\delta^{15}$N value (Fig. 5a) independently of the metamorphic grade. This decrease is also observed when N concentration is normalized to the concentration of fluid-immobile elements such as Al and Zr (Fig. 5b and c). In low-strain metagabbros, with the exception of sample Vi-52, both $\delta^{15}$N values and Cu concentrations appear correlated with the Mg# (Fig. 3). The recognition that Cu concentrations and $\delta^{15}$N values of low-strain metagabbros are strongly correlated (Fig. 5d, $r^2 = 0.96$) indicates that N and Cu geochemical behaviors were controlled by a common process. Cu/Al and Cu/Zr molar ratio of low-strain metagabbros are also roughly correlated with $\delta^{15}$N (Fig. 5e and f). In contrast, mylonites and veins plot significantly below the trend defined by the low-strain metagabbros (Fig. 5d, e and f). Among all the elements analyzed (reported in Tables 1 and 2), Cu is the only one to show a significant correlation with $\delta^{15}$N.

### Table 2: Major and trace element concentrations in mylonites and veins.

<table>
<thead>
<tr>
<th>Samples</th>
<th>Vi–262M</th>
<th>Vi–389M</th>
<th>Vi–385M</th>
<th>Vi–385V</th>
<th>Vi–387V</th>
<th>Vi–79V</th>
<th>VS–14V</th>
</tr>
</thead>
<tbody>
<tr>
<td>SiO$_2$</td>
<td>42.94</td>
<td>43.06</td>
<td>47.19</td>
<td>44.81</td>
<td>54.07</td>
<td>46.09</td>
<td>52.45</td>
</tr>
<tr>
<td>Al$_2$O$_3$</td>
<td>13.09</td>
<td>12.93</td>
<td>11.71</td>
<td>13.01</td>
<td>5.15</td>
<td>8.83</td>
<td>6.65</td>
</tr>
<tr>
<td>Fe$_2$O$_3$</td>
<td>19.60</td>
<td>20.33</td>
<td>15.94</td>
<td>17.85</td>
<td>9.94</td>
<td>9.25</td>
<td>10.80</td>
</tr>
<tr>
<td>MnO</td>
<td>0.33</td>
<td>0.41</td>
<td>0.31</td>
<td>0.38</td>
<td>0.11</td>
<td>0.07</td>
<td>0.05</td>
</tr>
<tr>
<td>MgO</td>
<td>4.82</td>
<td>4.94</td>
<td>5.46</td>
<td>5.06</td>
<td>9.22</td>
<td>7.64</td>
<td>7.71</td>
</tr>
<tr>
<td>CaO</td>
<td>10.27</td>
<td>8.17</td>
<td>9.49</td>
<td>9.26</td>
<td>15.00</td>
<td>14.58</td>
<td>14.19</td>
</tr>
<tr>
<td>Na$_2$O</td>
<td>4.32</td>
<td>4.12</td>
<td>5.21</td>
<td>4.73</td>
<td>5.86</td>
<td>5.19</td>
<td>7.05</td>
</tr>
<tr>
<td>K$_2$O</td>
<td>0.03</td>
<td>–</td>
<td>0.20</td>
<td>0.30</td>
<td>–</td>
<td>0.04</td>
<td>–</td>
</tr>
<tr>
<td>TiO$_2$</td>
<td>2.99</td>
<td>6.67</td>
<td>4.53</td>
<td>4.30</td>
<td>0.07</td>
<td>2.81</td>
<td>0.64</td>
</tr>
<tr>
<td>P$_2$O$_5$</td>
<td>1.93</td>
<td>0.74</td>
<td>0.24</td>
<td>0.82</td>
<td>3.15</td>
<td>0.21</td>
<td>–</td>
</tr>
<tr>
<td>LOI</td>
<td>–</td>
<td>–</td>
<td>–</td>
<td>–</td>
<td>0.48</td>
<td>0.93</td>
<td>0.77</td>
</tr>
<tr>
<td>Total</td>
<td>99.39</td>
<td>100.35</td>
<td>100.00</td>
<td>99.99</td>
<td>99.90</td>
<td>98.54</td>
<td>98.73</td>
</tr>
</tbody>
</table>

Ce      | 39.2    | 23.1    | 11.2    | 22.1    | 0.2     | 42.3   | 33.0   |
Co      | 23.2    | 15.7    | 23.0    | 27.5    | 12.4    | 19.1   | 35.1   |
Cr      | 15.9    | 12.5    | 8.0     | 12.6    | 1141    | 743    | 1007   |
Cs      | –       | –       | –       | –       | –       | –      | –      |
Cu      | 7.8     | 8.7     | 9.2     | 8.6     | 9.3     | 27.8   | 11.1   |
Dy      | 16.3    | 14.7    | 8.1     | 11.8    | 0.5     | 7.1    | 3.0    |
Er      | 9.2     | 8.6     | 4.6     | 6.3     | 0.5     | 1.8    | 0.8    |
Eu      | 5.3     | 3.0     | 1.3     | 2.2     | 0.1     | 7.1    | 3.2    |
Ga      | 22.3    | 19.6    | 21.0    | 21.5    | 12.1    | 20.1   | 22.9   |
Gd      | 18.6    | 12.1    | 6.2     | 10.1    | 0.2     | 20.2   | 8.3    |
Ge      | 2.1     | 1.9     | 1.6     | 1.9     | 1.3     | 1.2    | 1.2    |
Hf      | 6.7     | 3.8     | 2.6     | 2.8     | 0.2     | 4.7    | 1.0    |
Ho      | 3.3     | 3.0     | 1.7     | 2.3     | 0.1     | 0.9    | 0.4    |
In      | 0.2     | 0.1     | 0.2     | 0.2     | 0.1     | 0.1    | 0.1    |
La      | 9.5     | 6.5     | 3.2     | 6.3     | 0.1     | 8.3    | 10.7   |
Lu      | 1.2     | 1.2     | 0.7     | 0.8     | 0.1     | 0.2    | 0.1    |
Mo      | 0.6     | 0.6     | –       | 0.6     | –       | 0.5    | 0.7    |
Nb      | 3.4     | 6.7     | 3.2     | 3.2     | 0.5     | 3.3    | 0.7    |
Nd      | 48.9    | 24.5    | 11.0    | 22.4    | 0.3     | 67.2   | 33.1   |
Ni      | 19.1    | 34.8    | 33.3    | 23.3    | 484     | 155    | 297    |
Pb      | 22.6    | –       | –       | –       | –       | –      | –      |
Pr      | 8.2     | 4.3     | 2.0     | 4.1     | –       | 10.1   | 6.1    |
Rb      | –       | 0.7     | 6.7     | 7.1     | –       | 1.1    | 2.3    |
Sm      | 17.0    | 8.3     | 3.8     | 7.2     | 0.1     | 24.9   | 9.8    |
Sn      | 1.5     | 1.6     | 1.5     | 1.5     | 1.9     | 1.3    | 1.2    |
Sr      | 217     | 77.8    | 55.5    | 56.8    | 12.2    | 198    | 59.7   |
Ta      | 0.3     | 0.6     | 0.3     | 0.3     | –       | 0.3    | 0.1    |
Tb      | 2.7     | 2.2     | 1.2     | 1.8     | 0.1     | 1.9    | 0.8    |
Th      | 0.1     | 0.1     | 0.1     | 0.1     | 0.1     | 0.1    | 0.1    |
Tm      | 1.3     | 1.2     | 0.7     | 0.9     | 0.1     | 0.2    | 0.1    |
U       | 0.1     | –       | 0.1     | 0.1     | 0.2     | –      | –      |
V       | 421     | 492     | 616     | 619     | 98.8    | 350    | 301    |
W       | 0.5     | 0.4     | 0.7     | 0.7     | 0.4     | 0.3    | 0.5    |
Y       | 97.5    | 88.8    | 46.5    | 70.6    | 3.5     | 23.8   | 9.3    |
Yb      | 8.1     | 8.1     | 4.6     | 5.6     | 0.6     | 1.1    | 0.5    |
Zn      | 132     | 135     | 123     | 130     | 111     | 108    | 119    |
Zr      | 359     | 149     | 93.3    | 120     | 6.9     | 196    | 37.7   |

Major and trace element concentrations are respectively in wt% and ppm.

$(-1^\text{oo} to +8^\text{oo})$ Halama et al., 2010.
5. DISCUSSION

5.1. Nitrogen speciation in Alpine metagabbros

The determination of N speciation in metagabbros is difficult because N occurs in trace amounts (ppm level) and no N-bearing phases can be readily identified (except for the few secondary phengites present in eclogitic veins Vi–385V and Vi–79V). In situ techniques such as electron microprobe (Beran et al., 1992), ion microprobe (Hashizume et al., 2000) or nuclear reaction analyses (Gallien et al., 2004; Orberger et al., 2005) so far cannot resolve concentrations below...
few hundreds of ppm. Consequently, in the present work, N speciation is assessed indirectly. A first observation is that N concentrations reported in this study represent N fixed/bound in minerals – not in fluid inclusions – because N was extracted from fine powders (<60 μm) initially degassed under vacuum at 450 °C (see Section 3). This is consistent with results of the fluid inclusion analysis performed in omphacite from Monviso metagabbros, which showed that

Table 3
Results of N analysis for undeformed and low-strain Alpine metagabbros.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Microstructure</th>
<th>Weight (mg)</th>
<th>N amount (nmol)</th>
<th>N content (ppm)</th>
<th>δ15N (‰)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Chenaillet</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Amphibolite facies Fe-metagabbros (hydrothermal)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>CH-129</td>
<td>Undeformed</td>
<td>39.75</td>
<td>34</td>
<td>11.9</td>
<td>3.8</td>
</tr>
<tr>
<td>CH-129</td>
<td></td>
<td>32.88</td>
<td>27</td>
<td>11.4</td>
<td>4.7</td>
</tr>
<tr>
<td>CH-80-02</td>
<td>Undeformed</td>
<td>10.64</td>
<td>20</td>
<td>26.1</td>
<td>1.5</td>
</tr>
<tr>
<td>CH-80-02</td>
<td></td>
<td>28.94</td>
<td>62</td>
<td>29.9</td>
<td>1.2</td>
</tr>
<tr>
<td>Avg.</td>
<td></td>
<td>28.0</td>
<td></td>
<td></td>
<td>1.3</td>
</tr>
<tr>
<td>Queyras</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Omphacite</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>CH-129</td>
<td>Undeformed</td>
<td>39.75</td>
<td>34</td>
<td>11.9</td>
<td>3.8</td>
</tr>
<tr>
<td>CH-129</td>
<td></td>
<td>32.88</td>
<td>27</td>
<td>11.4</td>
<td>4.7</td>
</tr>
<tr>
<td>CH-80-02</td>
<td>Undeformed</td>
<td>10.64</td>
<td>20</td>
<td>26.1</td>
<td>1.5</td>
</tr>
<tr>
<td>CH-80-02</td>
<td></td>
<td>28.94</td>
<td>62</td>
<td>29.9</td>
<td>1.2</td>
</tr>
<tr>
<td>Avg.</td>
<td></td>
<td>28.0</td>
<td></td>
<td></td>
<td>1.3</td>
</tr>
</tbody>
</table>

Table 4
Results of N analysis for mylonites and veins from the Monviso massif.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Microstructure</th>
<th>Weight (mg)</th>
<th>N amount (nmol)</th>
<th>N content (ppm)</th>
<th>δ15N (‰)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Vi–262M</td>
<td>Mylonite</td>
<td>34.36</td>
<td>16</td>
<td>6.2</td>
<td>1.7</td>
</tr>
<tr>
<td>Vi–262M</td>
<td></td>
<td>54.34</td>
<td>26</td>
<td>6.5</td>
<td>1.7</td>
</tr>
<tr>
<td>Avg.</td>
<td></td>
<td>44.75</td>
<td>21</td>
<td>6.3</td>
<td>1.7</td>
</tr>
<tr>
<td>Vi–389M</td>
<td>Mylonite</td>
<td>67.56</td>
<td>52</td>
<td>10.6</td>
<td>0.8</td>
</tr>
<tr>
<td>Vi–385M</td>
<td>Mylonite</td>
<td>58.38</td>
<td>75</td>
<td>17.8</td>
<td>2.0</td>
</tr>
<tr>
<td>Vi–385V</td>
<td>Vein</td>
<td>29.00</td>
<td>40</td>
<td>19.0</td>
<td>2.0</td>
</tr>
<tr>
<td>Vi–385V</td>
<td></td>
<td>39.91</td>
<td>53</td>
<td>18.5</td>
<td>1.7</td>
</tr>
<tr>
<td>Avg.</td>
<td></td>
<td>39.46</td>
<td>46.7</td>
<td>18.7</td>
<td>1.9</td>
</tr>
<tr>
<td>Vi–387V</td>
<td>Vein</td>
<td>36.13</td>
<td>10</td>
<td>3.8</td>
<td>5.2</td>
</tr>
<tr>
<td>Vi–387V</td>
<td></td>
<td>40.49</td>
<td>12</td>
<td>4.0</td>
<td>6.5</td>
</tr>
<tr>
<td>Avg.</td>
<td></td>
<td>38.32</td>
<td>11</td>
<td>3.9</td>
<td>5.9</td>
</tr>
<tr>
<td>Vi–79V</td>
<td>Vein</td>
<td>13.77</td>
<td>16</td>
<td>15.8</td>
<td>4.0</td>
</tr>
<tr>
<td>Vi–79V</td>
<td></td>
<td>20.07</td>
<td>23</td>
<td>15.9</td>
<td>2.9</td>
</tr>
<tr>
<td>Avg.</td>
<td></td>
<td>16.93</td>
<td>19.9</td>
<td>15.9</td>
<td>3.4</td>
</tr>
<tr>
<td>VS–14V</td>
<td>Vein</td>
<td>9.47</td>
<td>39</td>
<td>57.2</td>
<td>4.0</td>
</tr>
<tr>
<td>VS–14V</td>
<td></td>
<td>32.50</td>
<td>124</td>
<td>53.1</td>
<td>4.7</td>
</tr>
<tr>
<td>Avg.</td>
<td></td>
<td>18.48</td>
<td>70.6</td>
<td>55.1</td>
<td>4.4</td>
</tr>
</tbody>
</table>
the inclusion fluid is devoid of nitrogen (Philippot and Selverstone, 1991). Nitrogen bound in crustal rocks usually occurs as ammonium, NH$_4^+$ (e.g. Vedder, 1964, 1965; Yamamoto and Nakahira, 1966). Ammonium is substituting for K$^+$ in K-bearing minerals such as feldspars and micas but can also occur in plagioclase feldspars substituting for Ca and Na (Honma and Itihara, 1981). In Alpine metagabbros, K concentrations are particularly low, generally below the analytical detection limit (i.e. <0.05 wt%). Petrological study of our samples indicates that K-bearing minerals are virtually absent. Nitrogen in Alpine metagabbros is thus believed to reside as NH$_4^+$ in Ca–Na-bearing minerals. Low-strain metagabbros show a crude correlation between N content and Na$_2$O + CaO content, supporting this idea (Fig. 4).

For the two veins where phengite was identified and separated (Vi–385V and Vi–79V), infrared spectroscopy analysis shows that they contain large amounts of ammonium (>500 ppm in single grains). This indicates that N primarily occurs as NH$_4^+$ in these veins. In contrast, the vein labeled VS–14V does not contain any phengite although its bulk-rock N content is the highest measured (~55 ppm N), indicating that nitrogen is carried by another phase. The mineralogy of this sample is dominated by omphacite, with minor talc and garnet. Interestingly, it also contains the highest Na content analyzed, suggesting that N may reside as NH$_4^+$ in the Na$^+$ crystallographic site. In support of this statement, infrared spectroscopy analyses of paragonite (i.e. Na white mica) separates from metasediments have NH$_4^+$ concentrations up to 120 ppm, showing that...
Na-bearing minerals can contain significant amount of N as ammonium, in agreement with the study of Honma and Itihara (1981). Accordingly, available evidence is consistent with N occurring mainly as NH$_4$ in Ca–Na minerals in low-strain metagabbros and in phengite in some high-pressure veins. The occurrence of NH$_4$ in mafic and ultramafic rocks is in full agreement with the recent conclusions from high-pressure experiments on clinopyroxene (Watenphul et al., 2009, 2010) and a study on mantle peridotite xenoliths (Yokochi et al., 2009).

5.2. Significance of the relationship Cu–δ$^{15}$N

Low-strain metagabbros show an inverse linear correlation between Cu concentration and δ$^{15}$N (Fig. 5d). This correlation could be inherited from (i) magmatic crystallization and differentiation, (ii) oceanic hydrothermalism or (iii) Alpine subduction process. These three hypotheses are discussed in details below.

Two samples from the Chenaillet massif, which have not been buried in a subduction zone, define a correlation indistinguishable from the one defined by subducted metamorphic rocks (Fig. 5d). This observation implies that the Cu–δ$^{15}$N relationship was acquired before the subduction stage either during magmatic differentiation or oceanic hydrothermalism. This idea is also supported by the fact that the Cu–δ$^{15}$N correlation does not show any progressive trend with increasing metamorphic grade (from non-subducted to blueschist- and eclogite-facies rocks).

With regards to magmatic differentiation, while N isotope composition may reflect magma degassing or variation in N speciation related to melt oxygen fugacity, Cu content distribution might be the expression of sulfide saturation and crystallization. Indeed, copper usually behaves as an incompatible element during differentiation of basaltic magmas (Brügmann et al., 1993; Alirezaei and Cameron, 2002), thus being progressively concentrated in the residual melt (Hamlyn et al., 1985) except if the magma reaches sulfur saturation. In the latter case, Cu behaves compatibly with respect to sulfide, resulting in a residual melt that is progressively becoming more depleted with respect to Cu and Mg. A compilation of data for mid-ocean ridge basalts shows a positive correlation between Cu and MgO contents (Fig. 6a; data from PetDB database, http://www.petdb.org/), illustrating that significant amount of Cu is stored by sulfides precipitated during magmatic differentiation. In Fig. 6, Alpine metagabbros are compared to MORB data. Comparing MORB and gabbros may not be straightforward because the gabbros are cumulates whereas the lavas represent melt compositions affected by variable degrees of differentiation. Gabbros can contain intercumulus melt (liquid) or could be cumulates from which such differentiated melt has been excluded (e.g. Alt and Anderson, 1991). In the case of Alpine metagabbros, relationships between Cu and MgO (Figs. 3j and 6a) and between Zn/Cu and MgO (Fig. 6b) are outlined. The overlap with MORB data (same Figures) suggests their comparison is meaningful. In Fig. 6a, Alpine metagabbros plot in part on the correlation defined by MORB, but it is not clear whether sulfide saturation was reached or not. Zinc is another incompatible element, which, compared to copper, is poorly partitioned into sulfide. The residual melt displays a negative correlation between Zn and MgO (not shown) and thus a positive correlation of Cu/Zn ratio with MgO (see MORB compilation in Fig. 6b). Fig. 6b shows that low-strain metagabbros define a single trend (except for one sample) that is inverse relative to the correlation defined by MORB. This indicates that during gabbro crystallization, sulfide saturation was not reached. Only one metagabbro with the highest MgO (13.44 wt%) appears comparatively enriched in Cu compared to other samples and might reflect sulfide accumulation (Fig. 6b).
Magmatic differentiation is unlikely to account for the $\delta^{15}$N–Cu relationship for at least three other reasons:

1. The normalization of Cu concentration to fluid-immobile elements such as Al and Zr (for which variations of concentration are only related to magmatic processes) also show negative correlations with $\delta^{15}$N (Fig. 5e and f). This indicates that variations of Cu concentration are mostly controlled by secondary process (i.e. non magmatic). Contrast with Al and Zr, Cu was added or lost from the rock by an interaction with a fluid.

2. Based on available experiments, it can be inferred that nitrogen is likely dissolved as N$_2$ in basaltic melt for oxygen fugacities ranging between IW (Iron–Wustite) to Air (e.g. Libourel et al., 2003; Roskosz et al., 2006). However, these studies were undertaken in anhydrous conditions and the occurrence of NH$_3$ might actually be more significant than determined experimentally. For example, under high $f_{H_2}$, amine groups (e.g. NH$_3$, NH$_2$-) are formed and stabilized (Mysen et al., 2008; Mysen and Fogel, 2010), and significant N-content and isotopic variations could be produced if nitrogen partitioning between melt and sulfide (and/or co-precipitating Ca–Na minerals) was important, leading to a decrease of N concentration in the magma with increasing differentiation. However, to our knowledge, the presence of NH$_3$ in magma has never been observed. This assumption is also at odd with the overall enrichment in N with decreasing Mg# reported in Fig. 3i.

3. Large N-isotope fractionation could also be recorded in the residual melt if nitrogen partitioning between melt and sulfide (and/or co-precipitating Ca–Na minerals) was significant, in particular if nitrogen was stored as ammonium. This process would produce a decrease of the magma N concentration together with an increase of $\delta^{15}$N due to isotope exchange between reduced and oxidized N species of $<3\%_{oo}$ (e.g. the isotope fractionation factor $\zeta_{NH_3,N_2}$ is <1.0032 for temperature higher than 930 °C; Scalan, 1957; Richet et al., 1977; Hanschmann, 1981; Haendel et al., 1986). If we assume that magmatic N$_2$ had $\delta^{15}$N $\sim$ $-5\%_{oo}$ typical of mid-ocean ridge basalt values (Javoy and Pineau, 1991; Marty and Humbert, 1997; Marty and Zimmermann, 1999; Cartigny et al., 2001b), NH$_4^+$ should have $\delta^{15}$N values between $-5\%_{oo}$ and $-2\%_{oo}$. This result contrasts with our dataset since rock samples with low N content (that would represent magma with low NH$_4^+$ content and high N$_2$ content) all show $\delta^{15}$N $>+4\%_{oo}$ (Fig. 5a).

The scenario about magmatic differentiation can be regarded with few modifications assuming that N was degassed during progressive differentiation. A loss of N$_2$ from the melt by degassing could produce a shift of N isotope composition if Rayleigh distillation process was associated with an isotope exchange between exsolved N$_2$ and dissolved N$_2$ (about $-1.2\%_{oo}$; Cartigny et al., 2001b) or NH$_4^+$ (about $3\%_{oo}$; Scalan, 1957). A degassing of N$_2$ from the magma would progressively decrease N content and increase $\delta^{15}$N value of the residual N in the melt. Following this model, N content decrease and $\delta^{15}$N increase should be correlated with a decrease of Mg#, which tracks magma differentiation. Fig. 3k and I shows the opposite in low-strain metagabbros since $\delta^{15}$N decreases with decreasing Mg# and increasing N contents. In other words, the lowest $\delta^{15}$N values and highest N contents are observed for the most differentiated samples.

The only alternative that can account for the Cu–$\delta^{15}$N linear correlation recorded in low-strain gabbros is a hydrothermal origin. Because samples with the highest N contents also show the highest Cu content (and vice versa), we suggest that N and Cu behave similarly, being either added to or leached from the rock by the circulating fluid. Experimental studies and analyses of altered ocean crust both show that hydrothermal fluids with temperature >300 °C and moderate water–rock ratio can mobilize substantial amounts of Cu (e.g. Seewald and Seyfried, 1990; Heft et al., 2008). By inference, this is also compatible with the idea that N is a fluid-mobile element (e.g. Bebb, 1995). Given the lower amounts of Cu in our metagabbros compared to most fresh MORB (Fig. 6a), it is likely that Cu and N were rather mobilized and leached out of the rock rather than added. This is also evidenced by the decrease of N/Al and Cu/Al ratio with increasing $\delta^{15}$N (Fig. 5), because Al is an immobile element during fluid–rock interaction. Although not exclusive, the simplest scenario that can be considered is the formation of copper(II) complexes with ammonia in the fluid. Copper(II) complexes with ammonia can be written as Cu(NH$_3$)$_2^{2+}$, where x varies mainly from 1 to 4 (e.g. Bjerrum, 1941). The formation of Cu(NH$_3$)$_2^{2+}$ complexes can strongly increase the solubility of both Cu and N in hydrothermal fluids. While N$_2$ can also be present in hydrothermal systems (e.g. Brettell et al., 1988; Yardley et al., 1993), it is not considered further in the discussion because N$_2$ should not form any chemical complex with Cu. An important question is whether NH$_3$ can be stable under pH and P–T conditions attending hydrothermal conditions similar to those experienced by Alpine metagabbros. The transformation of NH$_4^+$ to NH$_3$ is an acid–base reaction written as:

\[
\text{NH}_4^+ \rightarrow \text{NH}_3 + \text{H}^+ 
\]

The pK$_A$ of this reaction is lower than 4.8 at temperature higher than 300 °C (Olofsson, 1975). For conditions typical of high-temperature hydrothermalism (i.e., a fluid with pH >5 and amphibolite facies PT conditions), the dominant N species in the fluid will be NH$_3$ and therefore NH$_4^+$ can be partitioned from the rock to the fluid. NH$_4^+$ can be immediately transformed to NH$_3$, and NH$_3$ can react with Cu to form Cu(NH$_3$)$_2^{2+}$. In this model, Cu solubility is enhanced by NH$_3$ concentration, which in turn tends to scavenge Cu from the rock. In order to test quantitatively this hypothesis, we have modeled the evolution of Cu and N concentrations together with $\delta^{15}$N during leaching by hydrothermal fluids. Choosing an initial $\delta^{15}$N value of $-5\%_{oo}$ for the reference gabbro prior to leaching would lead to initial Cu and N contents of >250 and 100 ppm, respectively. Such high Cu and N concentrations are not compatible with typical mantle-derived magma (e.g. for N see Marty, 1995; for Cu see Fig. 6a). For this reason, we have assumed that the sample CH80-02, which contains the highest N and Cu concentrations (28 ppm N and 73 ppm Cu) and the lowest $\delta^{15}$N value among low-strain metagabbros ($+1.3\%_{oo}$), is representative of initial conditions. These initial values are different from typical mantle-derived melt; N content
(28 ppm N) is about 10 times higher and $\delta^{15}$N (+1.3‰) is at least 3‰ higher than basaltic melts. The origin of these high N content and $\delta^{15}$N values is unclear but may arise from (i) a moderately-compatible behavior of N during gabbro crystallization, potentially with N isotope fractionation between crystal and melt, and/or from (ii) melt-assimilation of deep crustal material enriched in N by seawater–rock interaction. Using the metagabbro CH80-02 as a starting point, two different calculations were performed considering (1) a Rayleigh distillation model or (2) a batch equilibrium model. Similar models were used in previous studies to simulate N isotope fractionation during fluid release associated with subduction of metasediments (e.g. Bebout and Fogel, 1992; Mingram and Bräuer, 2001). In our first model, we assumed that N occurred as NH$_4^+$ (see Section 5.1) and was progressively transformed and lost as NH$_3$ in the fluid. Assuming that each small aliquot of NH$_3$ produced was immediately transferred out of the system, the effect of N loss from the rock on its stable isotope composition can be modeled as Rayleigh distillation. The Rayleigh distillation describes an exponential enrichment of $^{15}$N over $^{14}$N in the residual NH$_4^+$ in the rock that is calculated as:

$$\delta^{15}\text{NNH}_4 = \delta^{15}\text{NNH}_4^0 + 1000(a - 1) \ln F$$

where $\delta^{15}$NNH$_4^0$ and $\delta^{15}$NNH$_4$ represent the initial and residual N-isotope composition of the rock, F is the N fraction remaining in the rock and a is the fractionation factor between NH$_3$ in the fluid and NH$_4^+$. We assumed that the N isotope fractionation between NH$_3$ and Cu(NH$_3$)$_2^{2+}$ was negligible. Fig. 7a shows that N isotope fractionation between NH$_3^+$ remaining in the rock and NH$_3$ leaving the system produces an increase of $\delta^{15}$N value in the residual rock. However, in Cu–$\delta^{15}$N and N–$\delta^{15}$N diagrams, the predicted $\delta^{15}$NNH$_4^0$ increases exponentially with decreasing Cu and N contents. This contrasts with the present data, which rather shows a linear correlation in a Cu–$\delta^{15}$N diagram (Fig. 5d). No combination was able to produce a good fit to the data, whatever the initial conditions (Cu and N contents, $\delta^{15}$N) and $\alpha_{\text{NH}_3}$–$\text{NH}_4^+$, and x-values (i.e. in Cu(NH$_3$)$_2^{2+}$). Accordingly,

![Diagram](image-url)
the loss of Cu and N from rock to fluid unlikely results from a Rayleigh distillation process.

In a second model, we considered that the rock and hydrothermal fluid interacted and evolved in isotope equilibrium before the fluid was expelled out of the system. Such a chemical exchange implies that the rock interacted with a same fluid phase during each step. This contrasts with the Rayleigh distillation model where the rock composition progressively evolves while interacting with different fluid aliquots that equilibrate as the reaction proceeds. The equation describing this batch equilibrium process derives from a simple isotope mass balance where the initial N present in the rock is equal to the sum of the N remaining in the rock and in the fluid phase at the end of the reaction. The equation for batch equilibrium model can be written:

\[ \delta^{15}N_{\text{NH}_4} = \delta^{15}N_{\text{NH}_4}^0 + 1000 \times F \times (z - 1) \]

The results of the model are shown in Fig. 7b. The best fit of the data is obtained for \( \delta^{15}N_{\text{NH}_4}^0 \approx 1.007 \) and \( x \approx 2 \) giving the Cu(II) complex Cu(NH\(_3\))\(^2+\). The isotope fractionation factor \( \delta^{15}N_{\text{NH}_4} \) of \( \sim 1.007 \) corresponds to a temperature of \( \sim 500\,^\circ\text{C} \) (Scalan, 1957; Hanschmann, 1981), in good agreement with amphibolite facies conditions. Although the batch model gives a good fit on the Cu–\( \delta^{15}N \) diagram, the fit is not as good on the N–\( \delta^{15}N \) diagram (Fig. 7b). While initial Cu contents of the “fresh” gabbros were likely homogeneous, initial N contents may have been slightly variable (but with homogeneous \( \delta^{15}N \)), for instance if the abundances of Ca–Na minerals were variable. The variability in N contents of Alpine metagabbros (Fig. 7b) would thus reflect concentration variation of the protoliths (i.e. unaltered gabbros), then modified by hydrothermal loss. In contrast, Cu contents would mostly record the hydrothermal leaching step.

We conclude that our batch equilibrium model is compatible with the present data. We note that oceanic hydrothermalism is usually considered to occur in open system (i.e. at each step, a new fluid is equilibrated with progressively evolved rock), a Rayleigh distillation would thus appear more realistic than batch equilibrium. However, Rayleigh modeling is not able to reproduce the trend observed in Cu–\( \delta^{15}N \) diagram. Alpine metagabbros represent deep hydrothermalism, i.e. in the lower part of the oceanic crust, and thus likely experienced limited metamorphism with low water/rock ratio, contrasting with the upper section of the crust which may involve larger amount of fluids in open system. In order to determine if altered basaltic up from upper oceanic crust show the same kind of trends between Cu and \( \delta^{15}N \), we have examined data from Sites 1256 (Busigny et al., 2005a,b) and 801C (Li et al., 2007). No correlation was found for these rocks, suggesting that Cu and N geochronological data are limited to lower crustal rocks (i.e. gabbros). The lack of correlation between Cu and \( \delta^{15}N \) in altered basalts may reflect that Cu was not mobilized at low-T, contrasting with high-T process (e.g. Seewald and Seyfried, 1990).

### 5.3. Nitrogen and copper behavior in subducted metagabbros

Assuming that metagabbros from Chenaillet are representative analogs of the eclogitic metagabbros from Monviso, Nadeau et al. (1993) estimated the amount of water lost during eclogitization to be about 90%, which is in good agreement with results obtained from experimental data (Schmidt and Poli, 1998) and geochemical modeling (Dixon et al., 2002). The discussion presented above shows that the Cu–\( \delta^{15}N \) linear correlation observed in low-strain metagabbros was inherited from a pre-subduction process, likely during oceanic hydrothermal alteration (see Section 5.2). If N would have been lost from the rocks during subduction metamorphism, isotope fractionation would have modified the rocks N isotope signature and shift them out of the initial linear correlation. Yet the preservation of the Cu–\( \delta^{15}N \) correlation during subduction implies that N and Cu remained stored in low-strain metagabbros during metamorphism although eclogitization was accompanied by \( \sim 90\% \) fluid loss. This observation is unexpected since N and Cu are usually considered as fluid-mobile elements (e.g. Seewald and Seyfried, 1990; Bebout, 1995; Heft et al., 2008). The recognition that all mylonites and veins analyzed here are located below the Cu–\( \delta^{15}N \) correlation defined by low-strain metagabbros (Fig. 5d) suggests that deformation had an impact on the Cu and N geochemical behavior. This implies either that mylonites and veins have lost part of the original N and/or Cu present in low-strain rocks, or that their \( \delta^{15}N \) was modified during subduction-zone metamorphism. Nitrogen loss during mylonitization or veining is unlikely because N concentrations of mylonites and veins are either similar or higher than low-strain rocks (Fig. 5a). In addition, a release of N to metamorphic fluids would produce an increase in \( \delta^{15}N \) of the residual rock due to preferential partitioning of light N isotopes to the fluid phase (e.g. Haendel et al., 1986; Bebout and Fogel, 1992; Mingram and Bräuer, 2001), which is opposite to our results (Fig. 5d). On the contrary, the low Cu contents of both the veins and mylonites (<11 ppm except one at 27 ppm) compared to low-strain rocks (from 6.4 to 73.2 ppm Cu, average = 41 ± 21 ppm 2σ) indicates that part of the original Cu present in low-strain rocks was lost during mylonitization and veining. The third and last hypothesis to be considered is a modification of \( \delta^{15}N \) values due to an external source. If the Cu concentrations of veins and mylonites are considered similar to the original low-strain rocks, then an interaction with a N-bearing fluid depleted in heavy isotope is required. N\(_2\) or NH\(_3\)-bearing fluids released from dehydration of crustal rocks are systematically depleted in heavy isotope relative to their host rock, being sedimentary, mafic (gabbros, basalts) or ultramafic (serpentine) in origin (e.g. Haendel et al., 1986; Bebout and Fogel, 1992; Mingram and Bräuer, 2001). The high N content of vein VS–15V (55 ppm) compared to other eclogite facies samples (<28 ppm N) strongly suggests that a fraction of the N carried in the veins may have been brought to the rock by external fluids. The presence of K in two of the veins investigated further supports that an externally-derived fluid phase carrying K and N infiltrated Monviso eclogite facies rocks at some stage of their HP metamorphic history. The slight K- and N- enrichment may point towards a sedimentary source rather than a mafic or ultramafic origin. Considering that the veins and mylonites were affected by an external N-bearing fluid, it is interesting to note that their \( \delta^{15}N \) values show a large range of variation (from 0.8‰ to
to 5.9%o). This implies that either (1) several fluids with various d15N values infiltrated the metagabbros and imprinted their geochemical signatures, or more likely that (2) N in mylonites and veins represent a mixing between N initially present in the rocks before subduction (with variable d15N) and N derived from external fluids (with homogeneous and light d15N).

To summarize, mylonites and veins have experienced either some loss of Cu and/or metasomatism by external fluids bearing 15N-depleted nitrogen. The present data suggest that, during eclogitization, N and Cu were affected by dynamic deformation of the rocks while static recrystallization favored their preservation into mineral structures.

5.4. Implication for nitrogen recycling from Earth surface to the deep mantle

The present work is the first to provide N content and isotope data on a suite of metagabbros. These rocks are broadly representative of the oceanic crust formed at ridge axes in modern divergent margins, hydrothermally-altered by seawater-derived fluids and subducted in a cold slab environment typical of present day settings (Peacock, 1996). However, it must be noted that the western Alps Jurassic ophiolites are usually regarded as being more “Atlantic-type” crustal sections in which rocks such as gabbros and even serpentinites are present near the top of the ophiolitic section and likely more vulnerable to seawater–rock interactions (e.g. Lagabrielle and Cannat, 1990). The “Pacific-type” oceanic crust thought to be subducting into most of Earth’s modern subduction zones in contrast shows the preservation of the pillow basalt section, at the top of the crust, and the gabbroic section may experience degrees of seafloor alteration smaller than those experienced in the “Atlantic-type” settings. Thus, western Alps metagabbros may have experienced greater water–rock interaction, and thus greater loss of N during high temperature hydrothermalism, than typical lower crust buried in modern subduction zones. This question is difficult to solve since the gabbroic section of the Pacific oceanic crust is still poorly known. Only recently the first sequence of an intact oceanic crust has been drilled from the lava, through the sheeted dike complex, down to the gabbros (Wilson et al., 2006), but its N content and isotope composition has not been analyzed yet. In the following discussion, we will assume that Alpine metagabbros are representative of lower crustal gabbros buried into modern subduction zones and we will use the present data to estimate average N concentrations and isotope composition of subducting metagabbros. A previous study of the Monviso metagabbros estimated that low-strain rocks and mylonites represent ~98% of the subducting metagabbros, while only 2% of the metagabbros correspond to veins (Philippot and van Roermund, 1992). Because veins are negligible in proportion (2%), and their N concentrations are strongly variable, we have estimated average N content and isotope composition based on low-strain rocks and mylonites only, that is 10.6 ± 5.1 ppm and +1.8 ± 0.8%o (1σ, n = 11), respectively.

The fluxes of N input in subduction zones by the various components of the oceanic crust are listed in Table 5. The flux of N carried by serpentinized peridotites is so far neglected because the mass flux of subducted serpentinites is presently underconstrained (see Philippot et al., 2007). Metagabbros are a major component of the subducted oceanic crust, with a flux of ~4 × 1016 g/yr (Peacock, 1990), twice that of basaltic crust. Considering the mean N content of mylonites and low-strain metagabbros calculated in this study, the flux of N subducted by metagabbros is estimated at 4.2 ± 2.0 × 1013 g/yr. We acknowledge that this estimate is based on a limited amount of data and that further study might lead to slightly different budget but first-order conclusions can be drawn. This value is half

Table 5
Nitrogen fluxes exchanged between internal and external reservoirs of the Earth.

<table>
<thead>
<tr>
<th>Input fluxes</th>
<th>Total flux (g/yr)</th>
<th>[N] (ppm)</th>
<th>1σ</th>
<th>N flux (g/yr)</th>
<th>δ15N (%o)</th>
<th>1σ</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sediments</td>
<td>1.8 × 1015a</td>
<td>424</td>
<td>±33</td>
<td>7.6 × 1011</td>
<td>+4.1f</td>
<td>±1.6</td>
</tr>
<tr>
<td>Upper crust (basalts)</td>
<td>2.0 × 1016b</td>
<td>6.5</td>
<td>±2.6</td>
<td>1.3 × 1011</td>
<td>+4.7</td>
<td>±2.4</td>
</tr>
<tr>
<td>Lower crust (gabbros)</td>
<td>4.0 × 1016b</td>
<td>5.9</td>
<td>±3.0</td>
<td>4.2 × 1011</td>
<td>+1.8</td>
<td>±0.8</td>
</tr>
<tr>
<td>Total input flux</td>
<td>13.2 × 1011</td>
<td></td>
<td>+3.4</td>
<td></td>
<td></td>
<td>+1.4</td>
</tr>
<tr>
<td>Output fluxes</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Mid-ocean ridges</td>
<td>–</td>
<td>–</td>
<td>–</td>
<td>0.7 × 1011c</td>
<td>–5.0</td>
<td>±2.0</td>
</tr>
<tr>
<td>Intraplate volcanism</td>
<td>–</td>
<td>–</td>
<td>–</td>
<td>5.7 × 107d</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>Arc volcanism</td>
<td>–</td>
<td>–</td>
<td>–</td>
<td>2.8 × 1011e</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>Back-arc basins</td>
<td>–</td>
<td>–</td>
<td>–</td>
<td>7.8 × 1094</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>Total output flux</td>
<td>3.6 × 1011</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

For input fluxes, we provide total mass flux, average N concentration, flux of N subducted and average δ15N for each component of the subducting oceanic crust.

c Estimated from N/He/Ar, C/N, mantle C-fluxes and C/Nb systematics after Cartigny et al. (2008) and data from Marty (1995), Javoy and Pineau (1991), Saal et al. (2002).
d Sano et al. (2001).
e Hilton et al. (2002).
f Estimate from a personal compilation (see Chapter II in Busigny, 2004).
the sedimentary N flux input in subduction zones estimated for comparable geotherms (7.6 × 10^{11} g/yr; Busigny et al., 2003a), showing that gabbros represent a significant source of N recycling to the mantle. The contribution of the basaltic portion to the deeply subducted lithosphere can be estimated from data on basaltic eclogites. Eclogites from Raspas Complex (Ecuador), with elevated δ^{18}O (Halama et al., 2011), and from Lago di Cignana (Italy), with relic pillow structure (van der Klaauw et al., 1997; Reinecke, 1998), are identified as metabasalts. These eclogites have a relatively restricted range of N content and δ^{15}N, with an average of 6.5 ± 2.6 ppm and +4.7 ± 2.4%_{\text{N}}^{\text{o}}, respectively (1σ, n = 19, excepting one sample with particularly high N content; Halama et al., 2010). Considering the mass flux of the upper basaltic crust buried in subduction zones is ~2 × 10^{16} g/yr, the flux of N carried by the upper oceanic crust is about 1.3 ±(0.5) × 10^{11} g/yr.

Fluxes input to subduction zones are compared to the fluxes of N output from the mantle by mid-ocean ridges, intraplate volcanism, arc volcanism and back-arc basins in Table 5 (see Section 1 for a detailed description of the fluxes). The total flux of N input in subduction zones (13.2 × 10^{11} g/yr) is almost five times higher than the flux of N output from arc volcanism (2.8 × 10^{11} g/yr; Hilton et al., 2002). Thus, about 80% of the N input is transferred to the mantle and not recycled to the surface (Table 5). This result is in good agreement with recent findings that N is massively cycled to the deep mantle in subduction zone environment (Busigny et al., 2003a,2005a,b; Sadofsky and Bebout, 2003; Li and Bebout, 2005; Li et al., 2007; Halama et al., 2010; Mitchell et al., 2010) and contrasts with early conclusions from the study of Central America subduction zones, in which a biased subduction budget was considered (Fischer et al., 2002; for a complete subduction budget see Li and Bebout, 2005; Li et al., 2007). A more recent study of volcanic gases in the Central America recognized the contribution of altered oceanic crust and proposed that N output flux from the arc may represent only 42% of the total input flux (Elkins et al., 2006). Accordingly, all detailed and recent studies confirm that a large portion of the N input in subduction zones is transferred to the deep mantle. When compared to the total flux of N output from the mantle, including mid-ocean ridges, arc volcanism, back-arc basins and intraplate volcanism (3.6 × 10^{11} g/yr), the total flux of N input in subduction zone is still significantly higher (more than three times higher). This implies that N contained in Earth surface reservoirs (mainly atmosphere but also hydrosphere, biosphere and crust) is progressively transferred and sequestered into the mantle, with a net flux of N input at ~9.6 × 10^{11} g/yr. Assuming a constant flux through time, the total amount of N present in the atmosphere (3.98 × 10^{21} g) could be sequestered in ~4.15 Ga. This net flux of N input to the mantle is thus insufficient to change significantly the amount of atmospheric N over a period shorter than 500 Ma.

Our flux estimates can also be used to constrain the past N distribution between atmosphere and mantle reservoirs. In particular, we can test the proposal of a recent study by Goldblatt et al. (2009), suggesting that (i) early Earth atmosphere contained twice as much N as today and (ii) during the first 2 Ga of the Earth history, half of this initial atmospheric N has been sequestered into the silicate Earth by way of subduction. This model was proposed to solve the so-called “faint young sun paradox”, i.e. the paradox of a non-permanent glaciated early Earth despite a poorly energetic young Sun. A major uncertainty in this model is that early subduction fluxes are largely unknown. Goldblatt et al. (2009) considered that the tectonic cycling was much more active, with a higher input flux than in modern subduction zones. However, they neglected that: (i) the flux of N output from volcanism was likely higher than the modern one (due to more active tectonics), (ii) the geothermal gradient in subduction zones was higher in the past than today and thus, a large amount of volatiles was likely released and not recycled to the deep mantle (i.e. low input flux; Bebout et al., 1999; Busigny et al., 2003a), and (iii) typical Archean marine sediments (i.e. Banded Iron Formations) contain a very small amount of N (few ppm; e.g. Beaumont and Robert, 1999; Pinti et al., 2001) relative to modern marine sediments buried in subduction zones (few hundreds of ppm). Although Archean shales contain similar N concentrations than modern sediments (e.g. Jia and Kerrich, 2004b; Garvin et al., 2009), they were likely deposited in “coastal settings” and it is unclear whether these were recycled into subduction zones. Overall, we can reasonably consider that the ancient flux of subducted N was similar to or smaller than the modern one. Assuming that early Earth atmosphere contained twice as much N as today (i.e. ~8 × 10^{21} g; Goldblatt et al., 2009), a constant flux of subducted N over the Earth’s history (~9.6 × 10^{11} g/yr) implies that only 1/4 of the early atmospheric N amount would have been sequestered into the silicate Earth over a period of 2 Ga. Accordingly, 50% of an early N-rich atmosphere cannot have been sequestered in the first 2 Ga of Earth’s history as suggested by Goldblatt et al. (2009), since it requires at least 4 Ga of subduction entrapment, even with the highest recycling fluxes considered above.

6. CONCLUSIONS

In the present work, we have examined typical oceanic gabbros, hydrothermalized by seawater-derived fluids, and then subducted along a low geothermal gradient (~8 °C/km). We find an inverse linear correlation between Cu and δ^{15}N for low-strain metababbros, which indicates, together with major and trace element data, that Cu and N geochemistry have been coupled during the formation and/or evolution of these gabbros. This correlation is incompatible with either a syn-subduction process or magmatic differentiation but rather derives from oceanic hydrothermal alteration. We predict that other altered gabbros would display such Cu–δ^{15}N relationship, while unaltered equivalents would not.

The preservation of Cu–δ^{15}N linear correlation for low-strain metababbros implies that static metamorphism associated to subduction did not affect significantly nitrogen and copper. Nitrogen and copper of subducting metababbros can thus be preserved to P–T conditions of 2.5 GPa and 500 °C. However, the study of mylonites and veins indicates that pervasive ductile deformation and veinling affected the
primary signal, which was associated with a release of Cu in fluids and/or a metasomatic addition of N by external fluids.

Considerations about global N budget confirm a large recycling to the deep mantle in subduction zones. Massive recycling may have imparted significant modifications of the N content and isotope composition in the Earth mantle and possibly atmosphere over geological time.

ACKNOWLEDGMENTS

Colleagues from the Isotope Geochemistry Laboratories in IPGP are thanked for fruitful discussions, particularly Jabrane Labidi, Jean-Louis Birck and Marc Javoy. Michel Girard, Jean-Jacques Bourrand and Guillaume Landais are acknowledged for their technical assistance. Marc Quintin is thanked for thin sections of all samples. Philippe Agard is thanked for discussions and providing a map from the Western Alps. Gray Bebout, Ralf Halama and Long Li are thanked for their careful and detailed reviews. Jeff Alt is greatly acknowledged for handling and reviewing the manuscript.

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Associate editor: Jeffrey C. Alt